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## Review Paper

# Review on Hyphenated Techniques in Pharmaceutical Analysis

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### ABSTRACT

The hyphenated technique is the combination or the coupling of the different analytical techniques. Mainly chromatographic techniques are combined with spectroscopic techniques. Then the separated components of the mixture from chromatographic technique will enter into the spectroscopic technique through an interphase. In GC-MS the separated components from gas chromatography enter to MS which is followed by ionization, mass analysis, and detection of mass-to-charge ratios of ions generated from each analyse by the mass spectrometer. Jet/orifice separator, effusion separator, and membrane separator can be used to connect GC with MS. In LC-NMR coupling the analytical flow cell was initially constructed for continuous-flow to NMR. However, the need for full structural assessment of novel natural products has led to the application in the stopped-flow mode in LC-MS. Use of LC-MS-MS is increasing speedily day by day. Hyphenated techniques such as HPLC coupled to UV and mass spectrometry (LC-UV-MS) have been extremely useful in combination with biological screening for a rapid survey of natural products. Nowadays, various types of LC-MS systems incorporating different types of interfaces are available commercially. The term hyphenated techniques refer to separation, identification, and the hyphenated techniques show better analysis of the samples are components specificity, accuracy, precision.

### INTRODUCTION

A hyphenated technique is combination or coupling of two different analytical techniques with the help of proper interface. Mainly chromatographic techniques are combined with spectroscopic techniques [1]. In the chromatography, the pure or nearly pure fractions

of chemical components in a mixture was separated and spectroscopy produces selective information for identification using standards or library spectra. The coupling of the separation technique and an on-line spectroscopic detection technology will lead to a hyphenated technique [2]. The term hyphenated techniques range from the combination of separation-seperation,

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separation-identification & identification-identification techniques[3].

The term “hyphenation” was first adapted by Hirsch Feld in 1980 to describe a possible combination of two or more instrumental analytical methods in a single run (Hirschfeld, 1980). The aim of the coupling is to obtain an information-rich detection for both identification and quantification compared to that with a single analytical technique [2].

## TYPES OF HYPHENATED TECHNIQUES

### A. Double hyphenated techniques

- ❖ GC-MS
- ❖ LC-MS

- ❖ LC-NMR
- ❖ LC-IR
- ❖ CE-MS
- ❖ GC-IR
- ❖ GC-NMR
- ❖ TLC-MS
- ❖ HPLC-DAD
- ❖ GC-FTIR

### B. Triple hyphenated techniques

- ❖ LC-MS-MS
- ❖ GC-MS-MS
- ❖ GC-IR-MS

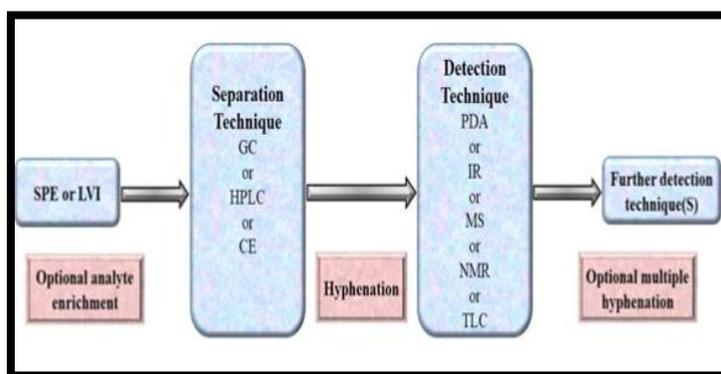


Figure:1 Schematic presentation of Hyphenation of chromatographic and spectrometric techniques

### A. Double hyphenated techniques

#### 1. GC-MS (GAS CHROMATOGRAPHY-MASS SPECTROMETRY)

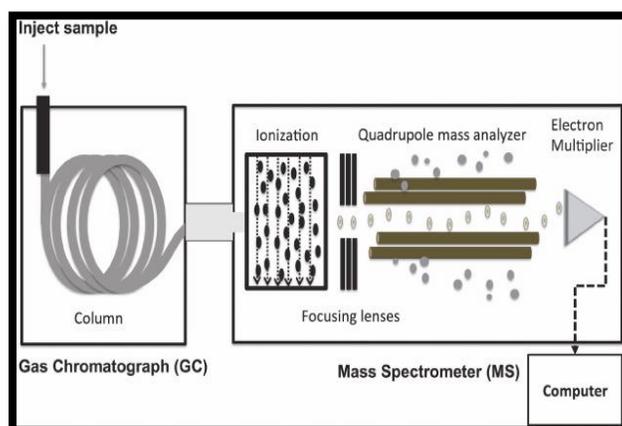
**PRINCIPLE:** The basis for gas chromatography/mass spectrometry (GC/MS) is the principle that heat causes a mixture to separate into its constituent substances. Prior to the analyte molecules being eluted into the MS for detection, the sample mixture is separated by the GC-MS instrument [4].

**INSTRUMENTATION:** Vaporized analytes undergo separation in the GC column when they are passed through it with the aid of heated carrier gas; this carrier is also referred to as the mobile

phase (helium). The chemicals separate as a result of interactions between the analyte, mobile phase, and stationary phase. The parameters of the column (length, diameter, and film thickness), the type of carrier gas, the gradient temperature of the column, and the characteristics of the stationary phase all affect the analyte’s separation. As the sample moves through the column, the mixture’s constituent parts separate due to variations in boiling points and other chemical characteristics. Because of their varied adsorption or variations in the partition between the mobile phases, the components have varying elution and retention times. After that, an interphase will allow the mixture’s separated components to enter the MS. Ionisation, mass analysis, and the determination of

the mass-to-charge ratios of the ions by each analysis by the mass spectrometer come next. GC and MS can be connected via an interface, such as a membrane separator, jet/orifice separator, or

effusion separator. Ionization is a process that separates a molecule into its positive and negative modes in addition to ionizing it [5].



**Figure 2: Schematic diagram of GC-MS**

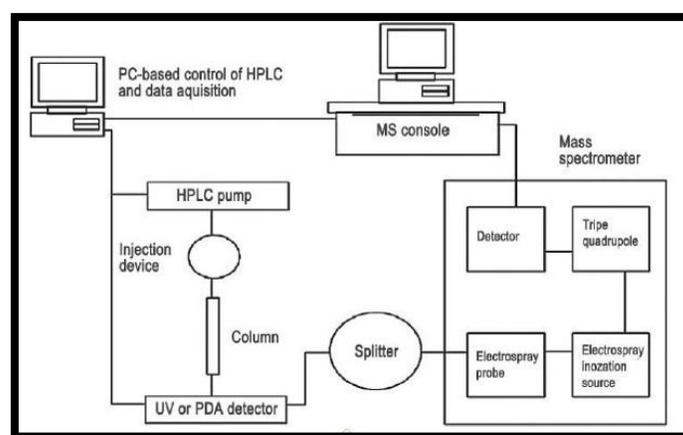
**APPLICATION:**

1. Gas chromatography having the highest resolution power to compare other techniques.
2. This technique having higher sensitivity when used with heated detectors [6].

**2. LC-MS (LIQUID CHROMATOGRAPHY – MASS SPECTROSCOPY)**

**PRINCIPLE:** The Liquid Chromatography-Mass Spectrometry (LC-MS) Principle Utilizing an HPLC, the LCMS technology first separates the constituent parts of a mixture before ionizing and separating the ions according to their mass/charge ratio[7].

**INSTRUMENTATION:**



**Figure 3: Schematic diagram of LC-MS**

Mass spectroscopy is combined with the physical separation of liquid chromatography, or HPLC, in a chemistry technique called LC-MS. A typical automated LC-MS system consists of a mass spectrometer, an LC system, and a double three-

way diverter connected to an auto sampler. The diverter usually serves as an automatic switching valve to direct unwanted portions of the eluting from the LC system to trash before the sample enters the MS. Soft ionization techniques are

generally used in LC-MS, which reveals mainly molecular ion species with few fragment ions.

A single LCMS run's worth of data is insufficient to confirm the molecule's identity. But the problem has now been resolved by tandem mass spectrometry (MS-MS), which distributes fragments by collision-induced dissociation of the generated molecular ions. The application of LC-MS-MS is expanding quickly every day. When combined with biological screening, hybridized techniques like HPLC coupled to UV and mass spectrometry (LCUV-MS) have proven to be incredibly beneficial for a quick survey of natural compounds [8].

#### APPLICATION:

1. Dissatisfaction with the expensive and inconsistent cross-reactivity of commercial immunoassays used in therapeutic drug testing has prompted the development of LC-MS assays as alternatives.
2. LC-MS analysis is pertinent for a number of steroid biochemistry domains. Due to

challenges in measuring low levels of testosterone and dihydrotestosterone seen in women and children, numerous highly sensitive LC-MS assays capable of delivering precise measurements in these classes have been developed[8].

### 3. LC-NMR (LIQUID CHROMATOGRAPHY-NUCLEAR MAGNETIC RESONANCE)

**PRINCIPLE:** LC separates components in a mixture based on their chemical properties. NMR provides detailed Structural information about the separated compounds. It involves a HPLC separation followed by the Detection of separated components by UV or other methods and ultimately NMR analysis. LC-NMR is a Powerful analytical tool used to resolving complex mixtures[9].

#### INSTRUMENTATION:

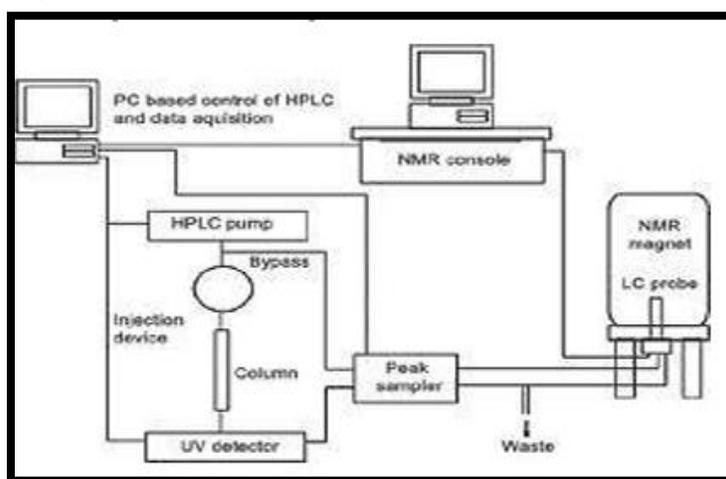


Figure 4: Schematic diagram of LC-NMR (5)

LC-NMR is a highly sensitive method, as its NMR sensitivity can be enhanced through the use of powerful magnetic field magnets and advanced sensitive probes. Additionally, improvements in peripheral technologies like solvent elimination techniques and automated measurement software

designed for multicomponent analysis contribute to this enhanced sensitivity. The greater the strength of the external magnetic field, the higher the sensitivity achieved[10]. The enhanced sensitivity has significantly shortened measurement times. Additionally, compounds

exhibiting intricate spectra can be readily analyzed due to the increased magnetic field, which enhances signal resolution[11].

#### APPLICATION:

1. Natural product analysis. Since then, a number of applications have been introduced to describe
2. Extracts from natural products.
3. Demonstrating products for drug degradation.
4. Impurities at moderate levels can be isolated and detected.
5. For environmental detection, this technique is utilized to track pesticides, herbicides and organic Contaminants.
6. Natural Products Chemistry: LC-NMR is widely used for the analysis of complex mixtures like Plant extracts, allowing researchers to identify and elucidate the structures of natural products [12], [13].

#### 4. LC-IR (LIQUID CHROMATOGRAPHY- INFRARED SPECTROSCOPY)

**PRINCIPLE:** The technique that combines liquid chromatography (LC) with infrared spectroscopy (IR) or Fourier-transform infrared spectroscopy (FTIR) is referred to as LC-IR or HPLC-IR. HPLC is regarded as one of the most effective separation methods available today, while IR and FTIR serve as valuable spectroscopic techniques for identifying organic compounds. This is due to the presence of numerous characteristic absorption bands in the mid-IR region that correspond to specific functional groups, such as –OH and –COOH. However, the integration of HPLC and IR presents challenges, and advancements in this combined technique have been notably slow. This is primarily because the numerous absorption bands from the mobile phase solvent in the mid-IR region can often mask the relatively weak signals produced by the sample components [14].

#### INSTRUMENTATION:

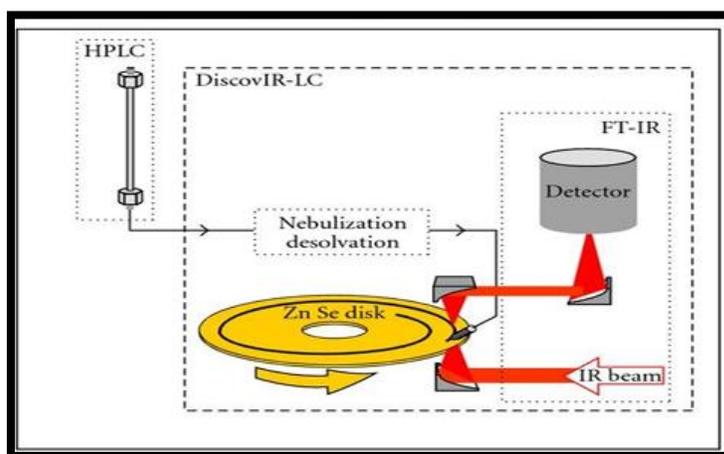


Figure 5: Schematic diagram of LC-IR

Every component ought to be able to be identified in real time with this device without chromatographic resolution being compromised. The most widely used method for this is mass spectrometry (MS), however it has some drawbacks, especially when it comes to

differentiating structural isomers like ortho-, meta-, and para-xylene, whose chemical-ionization mass spectra and electron-impact mass spectra are identical. A complementary technique to mass spectrometry is desired for such molecules. An alternative method for this purpose is Fourier

transform infrared (FT-IR) spectrometry, which produces unique spectra for the majority of structural isomers [14], [15].

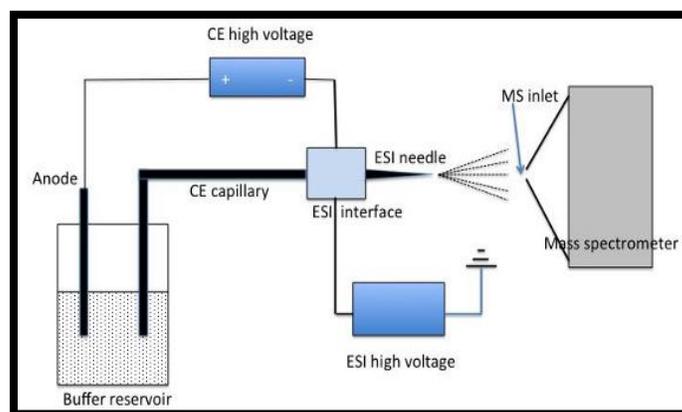
**APPLICATION:**

1. Eliminate the solvent without overflowing the vacuum system with diluent gas or thermally damaging the analyses.
2. Ensure that analyses are transmitted to the spectrometer efficiently.
3. Display the FT-IR in an analyte dense deposition.
4. Maintain the chromatography's resolution[16].

**5. CE-MS (CAPILLARY ELECTROPHORESIS-MASS SPECTROMETRY)**

**PRINCIPLE:** CE-MS is an analytical method. Mass spectrometry is gaining popularity as a detection technique for capillary electrophoresis (CE-MS). The integration of CE's superior efficiency with the high selectivity of mass spectrometry is highly appealing. CE is capable of handling complex sample matrices, making it an excellent pairing with mass spectrometry for analyzing a wide range of intricate mixtures. Additionally, mass spectrometry can enhance the overall sensitivity of CE analyses when conditions are suitable. One of the key advantages of combining mass spectrometry with any separation technique is the added dimension of separation it provides[17].

**INSTRUMENTATION:**



**Figure 6: Schematic diagram of CE-MS**

CE-MS is a hyphenated technique that links capillary electrophoresis (CE) and mass spectrometry (MS) using long capillaries. This connection can lead to increased analysis times, and there is also a shortage of suitable volatile buffers that are compatible with mass spectrometry[18].

**APPLICATION:**

1. for synthetic in vitro glycolysis studies.
2. for long term comparable assessment of the urinary metabolite.

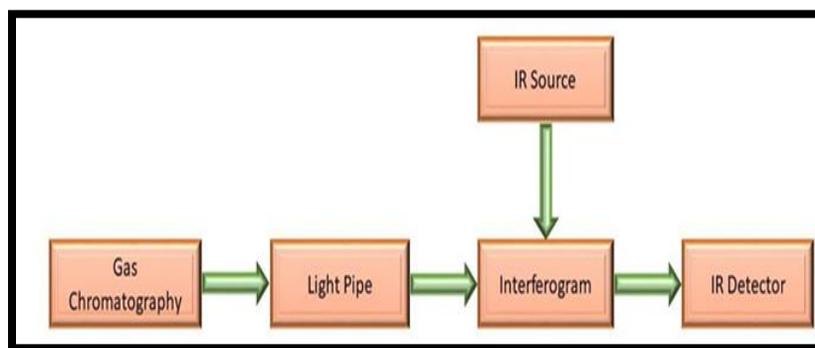
3. Characterization of monoclonal antibodies[10].
4. For Chiral separations.
5. for Forensic Analysis [19].
6. Analysis of O-Glycopeptides by Acetone Enrichment.
7. Identification of anthraquinone colouring matters in natural red dyes.
8. Determination of Drugs in Human Plasma.
9. Analysis of Inorganic Species [20], [21].

**6. GC-IR (GAS CHROMATOGRAPHY-INFRARED SPECTROSCOPY)**

**PRINCIPLE:** The technique involves the combination of gas chromatography and infrared spectroscopy. It is highly sensitive and quite costly. Additionally, sample recovery is achievable since infrared spectroscopy is a nondestructive method. In this process, gas chromatography handles the separation, while infrared spectroscopy perform the function identification [22], [23].

**INSTRUMENTATION:**

Gas chromatography separates the components of the analyte. These components will move through the column. These two techniques are connected via a glass column or vacuum tubes. The interface utilized in this method consists of a small glass pipe coated with gold on the inside, linked to the column by a narrow tube. The light pipe is heated to eliminate condensation and optimize the path length for improved sensitivity.



**Figure 7: Schematic diagram of GC-IR**

**APPLICATION:**

1. Pharmaceutical Industrial
2. DNA analysis of blood samples, other fluids[24].

**7. GC-NMR (GAS CHROMATOGRAPHY - NUCLEAR MAGNETIC RESONANCE)**

**PRINCIPLE:**

This technique involves the combination of gas chromatography (GC) and nuclear magnetic resonance (NMR). NMR is utilized to identify the components, while GC separates them. The integration of these methods offers structural information about the molecules of the separated components [25], [26].

**INSTRUMENTATION:**

The issue arising from combining these two techniques is that in NMR, the samples analyzed are either liquid or solid, while in GC, they exist in a gaseous state. When carrier gas is employed for

NMR analysis, it results in a low signal-to-noise ratio at atmospheric pressure. To address this sensitivity issue, microcells are utilized, along with computer systems, to enhance the signal-to-noise ratio [27]. Additional modifications are implemented, including the utilization of more powerful magnets and advanced microprobes [28]. Analytes with boiling points exceeding 65 °C condensed in the transfer capillary and probe head. This issue can be addressed by utilizing a transfer capillary that is heated by a bifilar coil. This coil is made from zero susceptibility wire and is combined with a strong magnetic field [29].

**APPLICATI**

**ON:**

1. Constitutional and Configurational isomers can be separated. Enantiomers shows the same spectra at different retention time.
2. An identification of stereoisomers in a complex mixture[29].

## 8. TLC-MS (THIN-LAYER CHROMATOGRAPHY-MASS SPECTROMETRY)

**PRINCIPLE:** The versatile instrument is used to isolate unknown compounds from a HPTLC/TLC plate and transfer them into a mass spectrometer for identification or structure elucidation.

### INSTRUMENTATION:

The TLC/MS interface can be integrated with any brand of LC-coupled mass spectrometer. It

features a plug-and-play installation utilizing two HPLC fittings within a standard HPLC-MS system. This semi-automatic instrument includes an automatic piston movement to create a pressure seal in the HPTLC/TLC zone on both glass plates and aluminum foils. Samples can be directly extracted from the plate using an appropriate solvent delivered by the HPLC/HPTLC pump, allowing for online transfer to the mass spectrometer. Additionally, the system includes automatic cleaning of the piston between extractions[30].

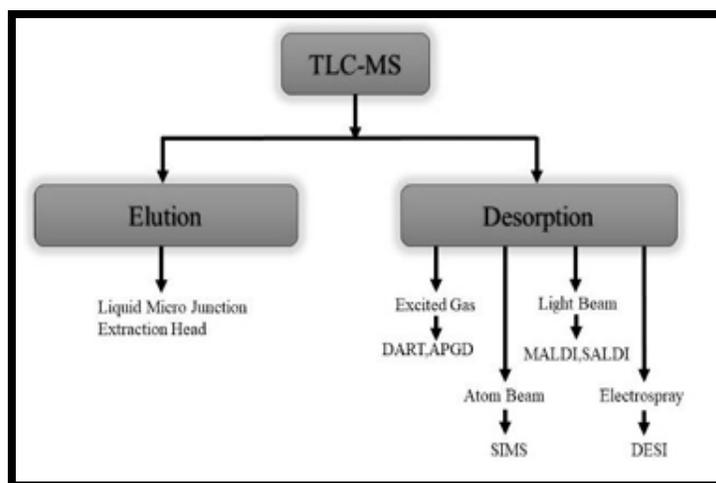


Figure 8: Schematic diagram of TLC-MS

### APPLICATION:

1. This method is undoubtedly the most crucial for ensuring sensitivity and rapid analysis.
2. Identification of active compounds in herbal and pharmaceutical products.
3. Analysis of cosmetics and environmental samples.
4. Clinical uses, such as studies on lipids and metabolites.
5. Identification of document forgeries, investigations into poisoning, and analysis of colorants.[31].

### B. Triple hyphenated techniques

## 1.LC/MS/MS (LIQUID CHROMATOGRAPHY-TANDEM MASS SPECTROMETRY)

**PRINCIPLE:** LC-MS (Mass Spectrometry) is capable of detecting over 300 compounds across various classes with a minimal injection volume. It requires minimal sample preparation, which helps to reduce analysis time. LCMS serves as the initial step in the LC-MS/MS process. This latter technique is significantly more sensitive and specific than LC-MS, offering sensitivity that is approximately 20 to 100 times greater. Its enhanced specificity is due to the incorporation of an additional filtering process during analysis[32].

### APPLICATION:

1. In forensic analysis, LC/MS/MS instruments are utilized to identify and measure the presence of drugs of abuse in biological samples like urine or blood.
2. These instruments are capable of detecting various drugs, including opioids, cocaine, and amphetamines[33].

## 2.GC/MS/MS (GAS CHROMATOGRAPHY TANDEM MASS SPECTROMETRY)

**PRINCIPLE:** This method combines gas chromatography with tandem mass spectrometry. It is both sensitive and specific, making it suitable for ultra-trace analysis. For qualitative identification using MS/MS, various scans can be employed, including product ion scans, precursor ion scans, and neutral loss scans with a triple quadrupole, as well as product scans using an ion trap. In recent years, the sensitivity of the quadrupole has improved, and its scanning speed has also increased[34].

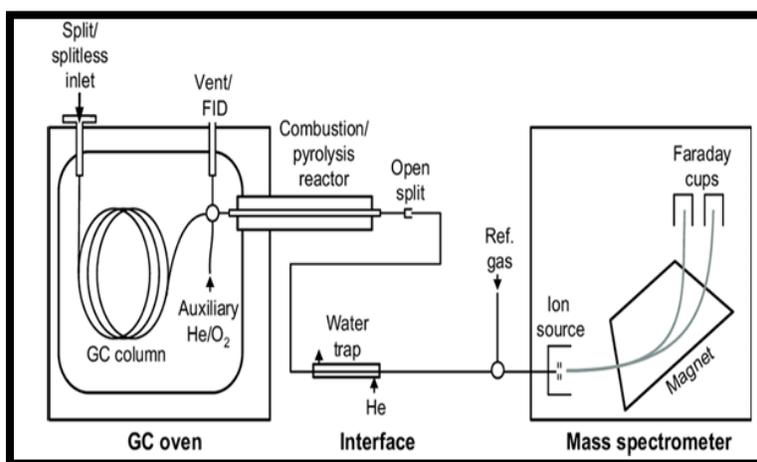
### APPLICATION:

1. Detection of trace unidentified impurities.
2. Utilized for assessing contaminations in environmental settings and food products, such as pesticides and PCBs found in food and biological samples[35].

## 3.GC/IR/MS (GAS CHROMATOGRAPHY–ISOTOPE RATIO MASS SPECTROMETRY)

**PRINCIPLE:** This principle has resulted in the widespread application of separation preparation systems coupled with mass spectrometers as detectors. This setup is commonly utilized to measure the stable isotope ratios of essential bioelements, such as carbon (C), nitrogen (N), oxygen (O), sulfur (S), and hydrogen (H). A gas chromatography-isotope ratio mass spectrometer (GC–IRMS) consists of four key components: the gas chromatograph, the high-temperature furnace, the water trap, and the isotope ratio mass spectrometer[36].

### INSTRUMENTATION:



**Figure 9: Schematic diagram of GC-IR-MS**

The ratios of the light stable isotopes of carbon (<sup>13</sup>C/<sup>12</sup>C), hydrogen (<sup>2</sup>H/<sup>1</sup>H), nitrogen (<sup>15</sup>N/<sup>14</sup>N), and oxygen (<sup>18</sup>O/<sup>16</sup>O) in individual chemical compounds are analyzed after separating them from complex mixtures. These isotope ratios in natural substances differ from those in synthetic

materials, allowing GC/C/IRMS to identify whether an analyte is synthetic. A key requirement is that the compounds in the sample mixture must be suitable for GC analysis, meaning they must be volatile and thermally stable. Some polar compounds may need additional chemical

modification or transformation (derivatization); in these instances, the stable isotope ratio of the derivatization agent must also be measured. Similar to GC/MS processes, the sample solution is injected into the GC inlet, where it is vaporized and transported onto a chromatographic column by a carrier gas, typically helium. As the sample moves through the column, the compounds in the mixture are separated based on their relative interactions with the column's coating (stationary phase) and the carrier gas (mobile phase). For the detection of carbon and nitrogen, the compounds exiting the chromatographic column enter a combustion reactor, which consists of an alumina tube filled with copper, nickel, and platinum wires, maintained at a temperature of 940 °C for oxidative combustion. This step is followed by a reduction reactor, also an alumina tube containing three copper wires kept at 600 °C, which converts any nitrogen oxides into nitrogen. A high-temperature thermal conversion reactor is needed for hydrogen and oxygen detection. Subsequently, water is removed using a water separator that directs the gas stream through a tube made from a water-permeable Nafion membrane, a sulfonated tetrafluoroethylene-based fluoropolymer-copolymer. Finally, the sample is introduced into the ion source of the mass spectrometer via an open split interface [37].

#### APPLICATION:

1. Assessing whether samples of chemically similar materials, including drugs, explosives, fibers, paints, inks, tapes, or adhesives, might originate from the same source or have a shared history.
2. Metabolic studies.
3. Sports medicine [37].

#### APPLICATION OF HYPHENATED TECHNIQUE IN ANALYSIS OF PHARMACEUTICALS

1. **Natural Product Analysis:** Hyphenated techniques like LC–NMR and LC–MS enable rapid structural analysis of natural compounds directly from crude plant extracts. They support dereplication and can be combined with online bioassays to study biological activities. On-flow and stop-flow LC–NMR help analyze unstable compounds, while at-line methods provide detailed structural information. Overall, these techniques offer fast and efficient natural product analysis. [38],[39]
2. **Structural Elucidation of Impurities:** In MS, pseudomolecular ions show molecular weight, but fragment ions provide structural information. In MS/MS, the selected ion is fragmented in a collision cell, producing daughter ions. The fragmentation pattern is used to identify the impurity structure. [40].
3. **Chemical Fingerprinting and Quality Control of Herbal Medicine:** The fingerprint technique is often used to isolate the matrix profile of a sample in the context of drug analysis. This is often enough to show the source and its preparation. For medicinal herbs, the profile depends not only on the production process, but also on the quality of the herbal raw materials. Conforms to renowned standards for assembly of structures. The integrated MS database also helps to identify these connections. Thus, GC-MS, LC-MS and MS-MS fingerprint profiles were obtained from the active ingredients of various herbal extracts, and the information was stored in the form of an electronic database that could be used for routine comparisons [41].
4. This approach enables the achievement of detection limits as low as the sub-picogram range in gas chromatography (GC) and femtogram range in liquid chromatography



(LC) [42]. Furthermore, along with enhanced quantification, the retention time data of target compounds facilitates qualitative analysis.

5. The integration of elemental mass spectrometry with HPLC was simpler than creating molecular mass spectrometric techniques that are coupled with HPLC, which followed shortly after the emergence of ICP-MS [43].
6. The coupling of HPLC and ICP-MS was also presented by Dean and colleagues. This integration was feasible due to the compatibility of HPLC separation flow rates with ICP nebulizers, allowing for the direct introduction of the HPLC eluent [44].
7. The integration of multiple hyphenated techniques enables the identification of new natural products, facilitating complete and definitive structure elucidation as well as the determination of relative configurations before engaging in the time-consuming and expensive processes of isolation and purification [44].42.
8. Isolation and analysis of natural products: Crude extracts of natural ingredients, which are very complex mixtures of many compounds, can be successfully analyzed using appropriate writing techniques. Among the various methods, LC-PDA and LCMS are the two most commonly used for the analysis of natural products. In addition to LC-NMR, various multiple binding techniques have recently been developed such as LC-PDA-NMRMS. If the ionization method is chosen correctly, LC-MS can be a very powerful and informative tool for the selection of crude plant extracts. The various types of LC-MS systems currently available allow the analysis of small non-polar compounds for large polar components such as oligosaccharides, proteins

and tannins, which are present in natural product extracts [45].

## ADVANTAGES OF HYPHENATED TECHNIQUES<sup>43</sup>

Hyphenated techniques combine multiple analytical methods to enable rapid identification of new natural products and detailed structure elucidation before time-consuming isolation and purification. A major advantage is their ability to detect unexpected chemical species, making them highly valuable in areas such as drinking water and wastewater analysis, drug discovery, biochemistry, and biotechnology.

1. Fast and accurate analysis.
2. Higher degree of automation.
3. Higher sample throughput.
4. Better reproducibility.
5. Reduction of contamination due to its closed system.
6. Separation and quantification achieved at same time.
7. Shorter analysis time.
8. Enhanced combined selectivity and therefore higher degree of information [46].

## CONCLUSION

Advances in the hyphenated technique such as LC-MS, GC-MS, LC-NMR, CE-MS and ICP-MS have been made to excellently solve various complex analytical problems in different fields. These techniques solve such problems in time efficient manner, higher degree of automation, higher sample throughput better responsibility. Combination of these techniques gives better analysis of the components. In this review introduction, instrumentation, and applications are explained for every technique.

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