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## **Review Article**

# Phytosomes: An Approach for Drug Delivery of Herbal Drugs

## Jyothi S\*, Nimisha Jain

Department of Pharmaceutics, Krupanidhi College of Pharmacy, Bangalore

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## **ABSTRACT**

Medicinal plants have long been valued for their therapeutic potential, offering a rich source of bioactive compounds such as flavonoids, alkaloids, and phenolic acids. Despite their pharmacological benefits, most phytoconstituents face significant barriers to effective clinical use due to their hydrophilic nature, poor lipid solubility, large molecular size, and limited ability to cross biological membranes, all of which contribute to poor bioavailability. To address these limitations, phytosome technology has emerged as an advanced drug delivery system. Phytosomes are molecular complexes formed by binding plant extracts or active phytochemicals with phospholipids, most commonly phosphatidylcholine, to create a lipid-compatible vesicular structure. This modification enhances solubility, stability, intestinal absorption, and overall therapeutic efficacy of plant-based compounds. Several preparation techniques, including solvent evaporation, rotary evaporation, ether injection, mechanical dispersion, lyophilization, salting out, antisolvent precipitation, and modern supercritical fluid methods, have been developed to optimize phytosome formation. Characterization approaches such as transmission and scanning electron microscopy (TEM/SEM), dynamic light scattering (DLS), differential scanning calorimetry (DSC), Fourier transform infrared spectroscopy (FTIR), X-ray diffraction (XRD), and nuclear magnetic resonance (NMR) provide essential insights into vesicle morphology, stability, drug entrapment, and molecular interactions. Phytosome technology offers multiple advantages, including improved bioavailability, reduced dosage requirements, enhanced therapeutic stability, and applications in both medical and cosmetic formulations. However, limitations such as rapid clearance of phytoconstituents remain a challenge. Overall, Phytosomes represent a promising bridge between traditional herbal medicine and modern pharmaceutical innovation, ensuring more effective clinical utilization of plant-derived therapeutics.

Address: Department of Pharmaceutics, Krupanidhi College of Pharmacy, Bangalore

Email : jyothireddy0801@gmail.com

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<sup>\*</sup>Corresponding Author: Jyothi S

#### INTRODUCTION

Since ancient times medicinal plants and their bioactive constituents have been traditionally used for centuries in the treatment of various diseases and health conditions.(1) Herbal medicine is an important alternative to synthetic drugs because it is often more affordable and generally associated with fewer side effects. Many countries still rely on traditional systems of medicine.(2) Plants produce a wide range of secondary metabolites, such as alkaloids, phenolic acids, flavonoids and terpenoids, many of which possess therapeutic properties. However, absorption challenges often limit bioavailability of these bioactive compounds.(3) Most active constituents in herbal medicines are hydrophilic in nature. Ecause of this property, their asorption inside the body or through topical application is often limited, leading to reduced therapeutic effectiveness. In addition, their relatively large molecular size restricts passive diffusion, and their low lipid solubility hinders movement across the lipid-rich membranes of poor bioavailability, making higher doses necessary to achieve the desired therapeutic effect.(4)

The term phytosome s derived from the word "Phyto" meaning "plant" and "some" meaning "cell" and it is sometimes also refered to as a herbosome.(5)(4) This technology was developed by the Italian company indena S.P.A to enhance the bioavailability of specific phytomedicines by incorporating phospholipids with standardised plant extract. Through this process the absorption and utilization of herbal compounds are significantly improved. The method involves forming stable lipid compatible molecular complexes between phospholipids and either standardized plant extracts or water soluble phytoconstituents thereby markedly increasing absorption and bioavailability.(4) Phytosome

technology provides a modern approach that improves not only the solubility of plant compounds but also their pharmacokinetic profile, leading to better absorption, distribution, and cellular uptake of secondary metabolites.(6)

## **BENEFITS OF PHYTOSOMES** (1)

- 1. **Better absorption:** when plant extract or bioactive compound combine with phospholipid, they show higher bioavailability due to improved absorption in the intestine.
- 2. **Cosmetic use:** Phytosomes are safe for cosmetic applications and help to improve skin penetration. Their lipid rich nature makes them suitable for transdermal delivery and functional cosmetics.
- 3. **Dose reduction:** better absorption allows effective results with smaller doses.
- 4. **Improved stability:** strong bonds between phosphatidylcholine
- 5. **As a carrier:** phosphatidylcholine in Phytosomes supports skin care and acts as a natural carrier.
- 6. **Low risk profile:** Phytosomal components are well studied and safe for large scale.
- 7. **Entrapment efficacy:** Drugs form stable vesicles with lipids, ensuring high and predictable entrapment.
- 8. Organic concentrates show greater bioavailability through phospholipid binding and enhanced retention in the digestive system.(4)

#### DISADVANTAGE OF PHYTOSOME

Fast removal of phytoconstituents can lower drug levels, showing the unstable of Phytosomes, which is key limitation.(1)

## STRUCTURE OF PHYTOSOMES



The extract structure of Phytosomes is a lipid compatible complex formed when bioactive plant constituents (usually water soluble) are chemically bound to phospholipid (choline head) binds to the plant compound, while the non-polar tails remain lipid soluble, creating a vesicle like arrangement that improves solubility, membrane passage and bioavailability.(5)

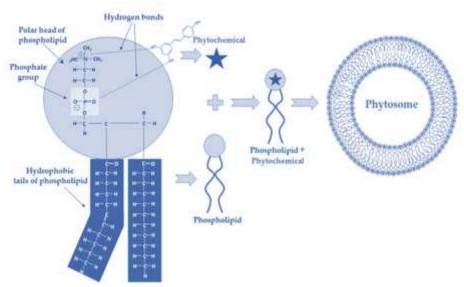


Figure 1. Schematic representation of the chemical bonds established during phytosome formation and their structure (7)

#### COMPONENTS OF PHYTOSOMES

## **Phospholipids**

Phospholipids can be broadly classified based on their backbone structure into sphingomyelins and glycerophospholipids. The latter group includes phosphatidylcholine (PC), phosphatidylethanolamine (PE), phosphatidylserine (PS), phosphatidic acid (PA), phosphatidylinositol (PI), phosphatidylglycerol (PG). Among these, PC, PE, and PS are the primary phospholipids used for forming complexes due to their characteristic structure comprising a hydrophilic head group and two hydrophobic hydrocarbon chains. Of these, phosphatidylcholine (PC) is the most widely utilized, owing to its amphipathic nature, which provides moderate solubility in both aqueous and environments. lipid Additionally, PC demonstrates excellent biocompatibility and low toxicity, reflecting its essential role in biological

membranes. Importantly, PC has also been associated with therapeutic benefits, particularly in the management of liver disorders, where it exhibits notable hepatoprotective effects.(8)

#### **Active constituents:**

The active plant constituents, many of which are polyphenols with poor solubility or limited membrane permeability. For instance, hesperidin is hydrophilic and cannot cross membranes, while rutin and curcumin are lipophilic but poorly soluble in gastrointestinal fluids. Phytosomes, polyphenols formed by complexing with phospholipids, enhance their ability to cross biological barriers, improving stability, absorption, and therapeutic effectiveness while protecting them from degradation by oxidation, hydrolysis, or photolysis.(9)

#### **Solvents**

The selection of solvents plays a crucial role in phytosome preparation, influencing phospholipid solubility, complex formation, and stability. Hydrophobic solvents (e.g., dioxane, dichloromethane, ethyl acetate, n-hexane, acetone) are commonly used, while ethanol and methanol

are preferred for food and pharmaceutical applications due to their low residue and ecofriendly nature, though their use is limited. A combination of protic and aprotic solvents, as well as supercritical fluids, has been shown to enhance solubility and efficiency in phytosome formulation. (10)

Table 1 Excipients used in phytosome formulation (9)

Chemical	Example	Uses
Solvents	methylene chloride, acetone	Aprotic solvent
Non-solvents	n-hexane or aliphatic hydrocarbons	Complex precipitation
Phospholipid	Egg phosphatidyl choline, soya phosphatidylcholine,	Cellular vesicles
	phosphatidyl choline, dipalmitoyl phosphatidylcholine	generating component
Alcohols	Methanol, ethanol	solvent
<b>Buffering agents</b>	7 % v/v ethanol tris buffer (pH 6.5) and Saline	Hydrating medium
	phosphate buffer (pH 6.5)	

## **METHODS OF PREPARATION** (11)

Table-2 methods of Phytosomes preparation

Method	Procedure	
Solvent	Plant extract and phospholipids are dissolved in acetone and refluxed at 50–60	
Evaporation	°C. The mixture is concentrated, filtered, and the dried precipitate (phytosome	
Method	complex) is stored in an amber glass container.	
<b>Rotary Evaporation</b>	Plant material and phospholipids are dissolved in tetrahydrofuran and stirred	
Technique	below 40 °C. After partial evaporation, n-hexane is added, and the resulting	
	precipitate is collected and preserved in an amber container.	
<b>Ether Injection</b>	The drug-lipid mixture dissolved in organic solvent is slowly injected into a	
Technique	heated aqueous phase, resulting in vesicle formation. Depending on amphiphile	
	concentration, structures may vary from micelles to vesicles.	
Mechanical	Phytoconstituents dissolved in diethyl ether are gradually added to an aqueous	
<b>Dispersion Method</b>	solution containing lipids. Removal of organic solvent under reduced pressure	
	yields the phytophospholipid complex. Supercritical fluid (SCF) approaches	
	(PCA, SAS, GAS) are modern alternatives.	
Lyophilization	Phospholipids and phytoconstituents are dissolved in suitable solvents, mixed,	
Technique	and stirred to form a complex. The complex is then isolated by freeze-drying	
	(lyophilization).	
Salting Out Method	Phosphatidylcholine and phytoconstituents are dissolved in aprotic solvents (e.g.,	
	acetone, dioxane) and stirred overnight. Addition of a non-solvent such as n-	
	hexane precipitates the complex, which is then separated.	
Antisolvent	Drug and phospholipids are refluxed in dichloromethane, concentrated, and then	
Precipitation	precipitated using hexane under continuous stirring. The precipitate is filtered,	
	dried, and sieved before storage.	
Supercritical Fluid	In GAS, the drug-phospholipid solution is mixed with a supercritical antisolvent	
(SCF) Technique	(e.g., CO <sub>2</sub> ) at controlled temperature and pressure. In SEDS, both solvent and	
	antisolvent are simultaneously introduced through a fine nozzle. These methods	
	yield high-purity phytosome complexes with improved stability and yield.	

## **Characterization of Phytosomes** (11)

**Table 3: Characterization Of Phytosomes** 

Technique	Description	
Visualization	Transmission electron microscopy (TEM) is employed to study vesicle size at high	
(TEM/SEM)	magnification (e.g., 1000×), while scanning electron microscopy (SEM) provides	
	details on particle morphology and surface characteristics. Dried samples are	
	mounted on gold-coated stubs for imaging.	
Particle Size &	Dynamic light scattering (DLS), supported by photon correlation spectroscopy	
Zeta Potential	(PCS), is used to determine vesicle size distribution and zeta potential, offering	
	insights into dispersion stability.	
Transition	Differential scanning calorimetry (DSC) measures the thermal transition of the	
Temperature	vesicular lipid system. Samples, including drug-phospholipid complexes and	
	controls, are heated between 0–400 °C at controlled rates in a nitrogen atmosphere.	
<b>Surface Tension</b>	The Du Nouy ring tensiometer (ring method) assesses surface tension in aqueous	
Measurement	drug solutions, reflecting amphiphilic behavior.	
Entrapment	Ultracentrifugation separates free drug from the phytosomal complex. The	
Efficiency	unentrapped drug concentration is quantified using UV spectroscopy to calculate	
	entrapment percentage.	
Vesicle Stability	Stability is monitored by evaluating vesicle size (via DLS) and structural integrity	
	(via TEM) at defined time intervals.	
Drug Content	The amount of incorporated drug can be quantified using validated analytical	
	techniques, including HPLC or UV–Vis spectroscopy.	

## **Evaluation of Phtosomes:**(9)

Ultraviolet Spectroscopy (UV): UV spectroscopy is employed to study the absorption behaviour of phytochemicals before and after complexation. Most reports indicate that phytosome formation does not significantly alter the absorption peaks of active components, suggesting that chromophoric groups remain intact.

Differential Scanning Calorimetry (DSC): DSC is used to examine thermal transitions such as melting points, peak shifts, and variations in enthalpy. Phyto phospholipid complexes usually show distinct thermograms compared to physical mixtures, indicating strong intermolecular interactions between phytoconstituents and phospholipid chains.

Fourier Transform Infrared Spectroscopy (FTIR): FTIR helps identify functional group interactions by analysing shifts in band position,

intensity, and shape. Comparison of complexes with pure compounds and their mixtures often reveals new peaks or altered intensities, confirming the formation of Phyto-phospholipid complexes.

X-ray Diffraction (XRD): XRD provides information on crystallinity versus amorphous character. The disappearance of sharp crystalline peaks in complexes compared to pure phytoconstituents suggests molecular dispersion and transformation into an amorphous form, which may enhance solubility and bioavailability.

**Nuclear Magnetic Resonance (NMR):** <sup>1</sup>H and <sup>13</sup>C NMR techniques allow for detailed structural analysis. Results indicate that phytosome formation is stabilized mainly by hydrogen bonding between polyphenolic groups and phospholipid head regions, rather than covalent bond formation.

### **CONCLUSION**



Phytosome technology represents a significant advancement in enhancing the therapeutic potential of herbal medicines by overcoming the of poor solubility limitations and bioavailability. By forming stable complexes between plant-derived bioactive compounds and phospholipids, Phytosomes improve absorption, stability, and pharmacokinetic behaviour, making them effective for both pharmaceutical and cosmetic applications. Various preparation techniques—ranging from solvent-based methods to modern supercritical fluid approaches enable the efficient development of these complexes, while advanced characterization tools such as TEM, SEM, DSC, FTIR, XRD, and NMR confirm their structural integrity and functionality. Collectively, these features establish Phytosomes as a promising drug delivery system that bridges traditional herbal medicine with modern therapeutic efficiency.

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