



**INTERNATIONAL JOURNAL OF
PHARMACEUTICAL SCIENCES**
[ISSN: 0975-4725; CODEN(USA):IJPS00]
Journal Homepage: <https://www.ijpsjournal.com>



Research Article

Henna Dyeing On Cotton Fabric With Different Natural Mordants And Henna Tattoo With Different Natural Solvents

Vibha S. Bhagat*, Afsana Mondal, Mahedaraksha Junaid Ahmed, Juliet Miranda

Department of Chemistry, Royal College, Penkarpada, Mira Road (E), Thane, 401107, India

ARTICLE INFO

Received: 24 June 2024

Accepted: 28 June 2024

Published: 17 July 2024

Keywords:

Henna dye, henna tattoo, fastness properties, mordants and solvents.

DOI:

10.5281/zenodo.12761219

ABSTRACT

Natural dyes have a lesser harmful effect on living bodies. Cotton is used as a fiber due to its higher ability of dye intake. To determine the darkness percentage of henna with different mordants. Henna powder is used for dyeing cotton with different mordants like water, lemon, sugar water, salt water and oil. Fastness Properties of washing, rubbing and light of all the samples were determined by grayscale. For henna tattoo, henna was applied with different solvents like water, oil, lemon and salt to determine the darkness percentage. Body temperature of all three persons were recorded. It showed different results with different people due to the biochemistry of different bodies.

INTRODUCTION

Various industries, in recent times, use dyes for printing textile, paper, leather and different materials for coloring their products[16]. This brings customers interest towards colorful shades of the product. Among all industries, textile is one of the main industries which use dye for coloring their fabric.[16] According to estimations, more than 100000 commercial dyes are known, with annual production of more than 700000 tonnes in a year[16]. However, Dyeing fabric with synthetic dyes has led to several worse effects not only to the water bodies through effluents but also to human health.[16] Major Synthetic dyes used are

azo dyes about 60% - 70% for dyeing. [1],[3] These azo dyes about 10% - 15% are released into the environment through effluent, as they don't bind to the fibre.[3] Synthetic dyes like cationic dyes, releases effluent with dye in it; which contains harmful heavy metals like Copper (Cu), Cobalt (Co), Zinc (Zn), Lead (Pb), Chromium (Cr) in aqueous form. Effluent of Anionic dyes have hydroxyl anions in it, which are soluble in water; hence it's difficult to remove anions from the effluent. [16] In most studies, they are considered carcinogenic to health [2],[3] and can cause several health hazards to the environment.[3],[7] They are toxic, mutagenic and carcinogenic.[3]

*Corresponding Author: Vibha S. Bhagat

Address: Department of Chemistry, Royal College, Penkarpada, Mira Road (E), Thane, 401107, India

Email ✉: vibhabhagat.royal@gmail.com

Relevant conflicts of interest/financial disclosures: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.



Though many of the dyes are banned, some of them are still allowed, only when used within a certain limit. [16] Whereas natural dyes are a safer choice as it has minimal impact on the environment.[16] They are derived from animals, plants, insects and minerals.[16] They are non - substantive and can be used for dyeing only along with mordant.[16] The substantive natural dye is turmeric which does not require any mordant and can be applied directly. [16] Lawsonia inermis L. Commonly called henna, is a large, evergreen shrub or small tree with a densely branched habit.[4] The staining property of henna is mainly attributed to a 2-hydroxy-1,4-naphthoquinone compound named as lawsone also known as hennotannic acid which is seen in abundance in the dried leaves. [4],[17] Henna dyeing when done without using any harmful mordants, leaves no hazard behind to the living bodies of the environment.[17] Hence the dyeing interest is now shifting towards the naturally occurring dyeing agents.[14],[17] Henna has lawsone pigment which gives red-orange color.[15] Henna is substantive dye for protein fiber and textile fibres. [17] Similarly, carrot, beetroot, blueberry, papaya, leaves, etc. Also have carotenoid, betalain, anthoxanthin, chlorophyll pigments which helps in different shades of colouration. There have been many researches on the dyeing ability of henna on cotton fabric. Which showed that the darkness of henna depends on temperature [7] and different percentage dilution with water, [8],[9] the type of solvent used like Hot and cold water and ethanol solvents, [10] different types of fiber, [10] the pH of the dye. [11] Different mordants used [10] with henna also gave different results. Another property that, the coloring compound of henna exhibit is, 'Lawsone' strongly absorbs UV light.[6] Hence, henna irradiated with UV radiation can show difference in color intensity.[7] For this, studies

showed that the dye extracted from the henna at an optimum temperature 65°C, with radiated henna depicted good staining on fabric.[7] The implications of henna dye have been shown to have poor to moderate dyeing capability towards cotton fabrics. [9] However, cotton has a higher rate of dye intake than other fiber, [10] hence, we used cotton in this study. The color fastness with respect to light exposure, washing and rubbing have been found satisfactory in previous papers. [12] Fastness properties of dyed woolen yarn samples were also found to be considerably good. [13] Mordants are the fixing compounds which are used to lock dyes to the fabric.[17] Those mordants are mostly metallic salt, which are considered to be hazardous to certain extent.[17] Hence, studies are focussing more on dyeing without mordant.[17] However, in this study, we used lemon, oil, salt water, sugar water as natural mordant to check the intensity of dyeing. Whether these natural mordants will have the dye to fabric binding capacity. For this, cotton was the fabric selected, as its better binding capacity with the dye.[10] Fastness properties like washing, light and rubbing of the samples were measured using Colors grab software application. Henna tattoo - Henna is used as a decorative material applied on hands, feet, nails, hair, etc.[19] Henna powder is extracted by drying, milling and sifting the leaves.[19] Traditionally, oil and sugar is used for extracting dye and deepening the color.[19] Sometimes, henna is adulterated with jagua (genipin and geniposide are the coloring components) and PPD (p-phenylenediamine) which intensifies the color.[19] Some studies showed that PPD is used with henna for blackening the color and has been found to be allergic.[20] Genipin was the main component that can cause allergic effect.[19] During ancient period, henna was used as medicine externally or



internally for jaundice, leprosy, smallpox, and skin complaints.[18],[19] Lawson coloring compound in henna reacts chemically with the protein keratin in skin resulting in a strong permanent stain that lasts until the skin is shed. [5] This could be because henna pierces the outer dead skin cells.[6] During review of literature, it was observed to lack studies in the field of henna tattoo. We are focusing on filling those gaps as there is little information regarding skin tattoos. Here, we used different solvents to identify the difference in intensities of three people. The darkness percentage was recorded by software application - color grab upto 96 hours.

METHODS AND PREPARATION

Preparation of henna samples-

1. Blank sample - 2gm henna Powder + 10ml warm water about 40-60°C
2. Lemon sample - 2 gm Henna Powder + 5ml lemon extract + 5 ml warm water
3. Sugar sample - 2 gm Henna Powder + % 10 ml sugar in warm water
4. Salt sample - 2 gm Henna Powder + % 10 ml salt in warm water
5. Oil sample - 2 gm Henna Powder + 5 ml coconut oil + 5 ml warm water

Methods used-

Cotton has a high affinity to dye. The white cotton cloth was scoured with salt (NaCl) and detergent supplied by Procter and gamble company and bleached using Sodium hypochlorite (NaClO) to remove undesirable or unwanted compounds attached to it. The cotton was then dipped into the evaporating dish containing the sample. Each sample had three cotton pieces labeled as A,B,C. First batch was washed after 30 minutes. Another batch was washed after 31 hours. The cloth pieces were washed thoroughly with detergent. The measurements of its darkness % from grayscale was taken down as reading from colors grab, a

software application. Washing Fastness property was recorded after washing with detergent and drying, pieces labeled as 'A'. Their darkness % from grayscale was recorded. Light fastness property was obtained keeping the pieces labeled as 'B' in sunlight for 3-4 hours. Their darkness % from grayscale was recorded. Rubbing fastness property was obtained by rubbing the pieces labeled as 'C'. Darkness % from grayscale was recorded.



Fig. No. 1. Prepared samples for dyeing

For henna tattoo –

Henna extract was supplied by ASHTANG healthcare pvt. Ltd.

4 samples were used

1. Blank - Henna Powder + water
2. Oil - Henna Powder + oil
3. Lemon - Henna Powder + lemon extract
4. Salt - Henna Powder + salt water

The water used here was lukewarm between 40°C - 60°C. Before applying henna, their body temperatures were recorded. Later, these henna samples were applied on skin thrice for 1 hour. The measurements were recorded from colors grab, a software application for consecutive 4 days after each 24 hrs including after removal reading.



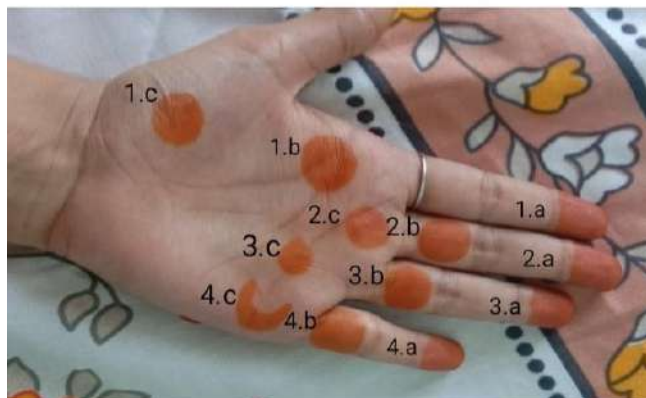


Fig. No. 2. Henna tattoo just after removal. (1. Blank, 2. Oil, 3. Lemon, 4. Salt)



Fig. No. 3. Samples after 30 mins. dyeing and washing



Fig. No. 4. Samples after 31 hrs dyeing and washing

Standardization of readings -

For fabric dyeing, external factors like light intensity, reflection, time difference, contact with water, different locations were the main cause of error in reading. Hence, for standardizing these conditions, pictures were clicked and analysed at night under the same light, time and location; to avoid error due to the difference in light intensity of sunlight. For henna tattoo, biochemical factors like hormones, body temperature, contact with water, light intensity, reflection and different locations caused error in reading. To avoid this, three tests for the 4 samples were simultaneously performed on one hand, which is in less contact with water. And to avoid hormonal changes and body temperature changes between two consecutive tests. Pictures were clicked and analysed at night under the same light, time and location to avoid light intensity differences between consecutive readings

OBSERVATIONS

For cotton dyeing-

Dyeing time : 30 minutes.

	Darkness 'A'	Washing	Darkness 'B'	Light	Darkness 'C'	Rubbing
Sample 1 (Blank)	24	24	22	23	21	21
Sample 2	24	23	24	23	25	24
Sample 3	23	23	22	22	25	23
Sample 4	21	20	24	22	22	20
Sample 5	27	25	26	24	29	27

Dyeing time : 31 hours.

	Darkness 'A'	Washing	Darkness 'B'	Light	Darkness 'C'	Rubbing
Sample 1 (Blank)	25	24	24	25	29	29
Sample 2	28	25	24	23	26	25
Sample 3	28	26	28	25	26	25
Sample 4	24	23	24	23	23	23
Sample 5	37	36	36	35	39	39

For henna tattoo -

Person 1 (Body temperature - 97.6 F)

	1 (blank)	2	3	4	1 (blank)	2	3	4	1 (blank)	2	3	4
0 min	32	32	41	35	38	37	43	32	51	42	46	44
24 hrs	36	38	45	40	42	41	55	50	35	26	35	35
48 hrs	23	27	32	28	35	25	37	42	25	20	21	33
72 hrs	22	24	28	27	31	23	28	35	24	22	31	29
96 hrs	19	18	26	26	24	21	25	25	22	11	28	19

Person 2 (Body temperature - 97.9 F)

	1 (blank)	2	3	4	1 (blank)	2	3	4	1 (blank)	2	3	4
0 min	48	39	34	45	41	40	43	39	47	35	32	33
24 hrs	56	50	51	62	54	49	49	49	50	54	47	47
48 hrs	56	52	58	59	54	52	52	52	57	56	58	53
72 hrs	45	41	51	45	43	49	45	44	49	55	56	52
96 hrs	33	35	41	36	42	42	35	35	38	36	41	38

Person 3 (Body temperature - 96.3 F)

	1 (blank)	2	3	4	1 (blank)	2	3	4	1 (blank)	2	3	4
0 min	60	54	57	65	57	57	64	60	56	54	56	56
24 hrs	53	52	51	52	51	48	50	49	52	46	50	49
48 hrs	41	29	40	48	42	41	38	38	43	45	37	46
72 hrs	34	29	42	29	32	38	36	31	42	40	36	41
96 hrs	30	28	34	27	30	32	38	31	34	38	35	29



OBSERVATIONS

Cotton dyeing-



Fig. No. 3. Washing fastness (wet) samples 'A' (upper samples 30 mins. And lower samples 31 hrs)

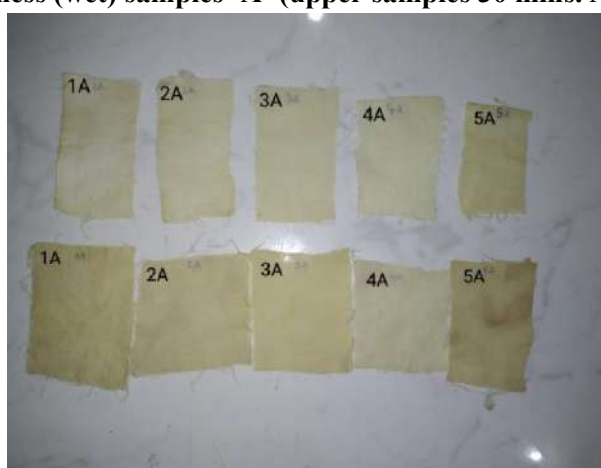


Fig. No. 4. Washing fastness property (dried) samples 'A' (upper samples 30 mins. And lower samples 31 hrs)

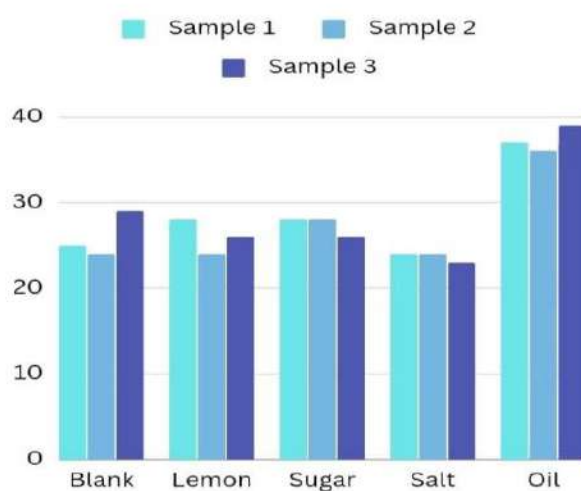


Fig. No. 5. Light fastness property of samples 'B' (upper samples 30 mins. And lower samples 31 hrs)

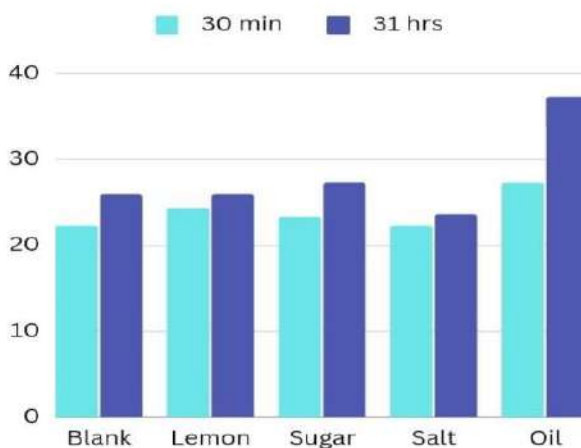


Fig. No. 6. Rubbing fastness property of samples 'C' (upper samples 30 mins. And lower samples 31 hrs)

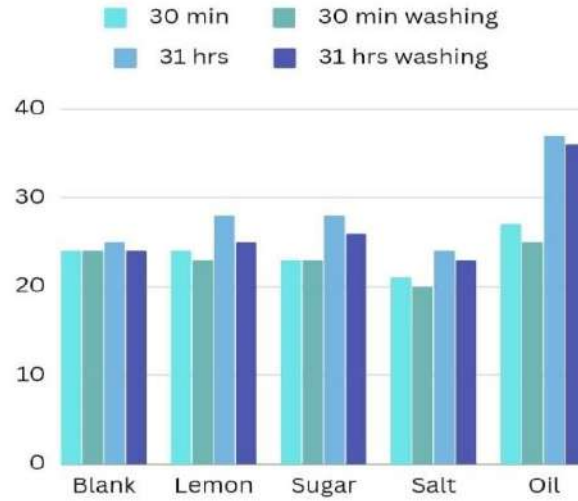
31 hrs graph -



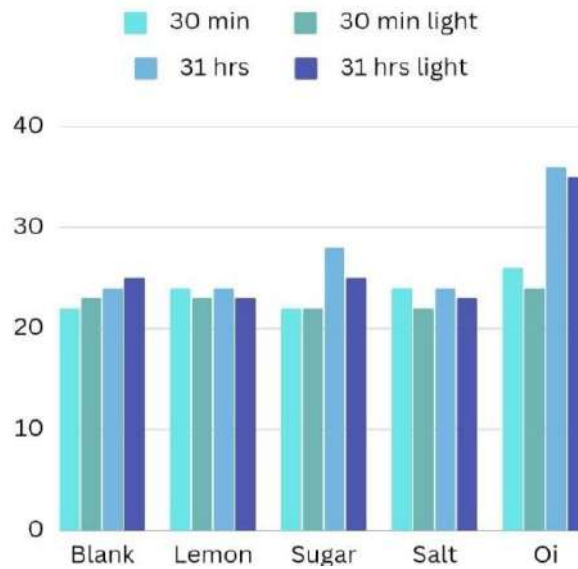
Comparison of 30 min and 31 hrs dyeing-



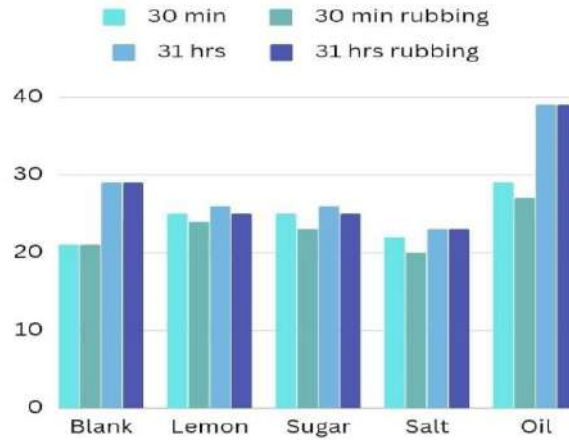
Comparison before and after washing



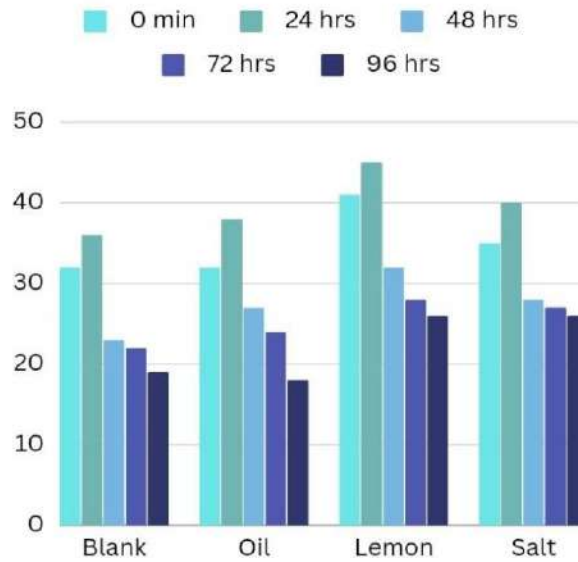
Comparison before and after light exposure



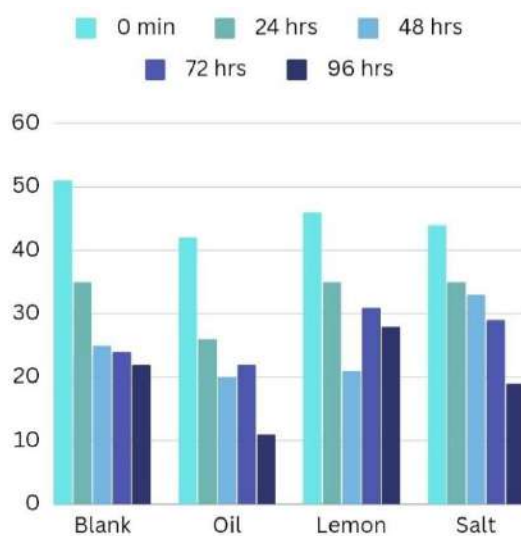
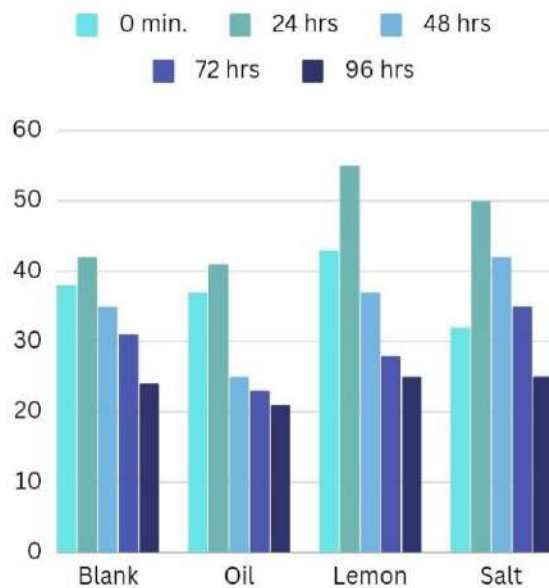
Comparison before and after rubbing -



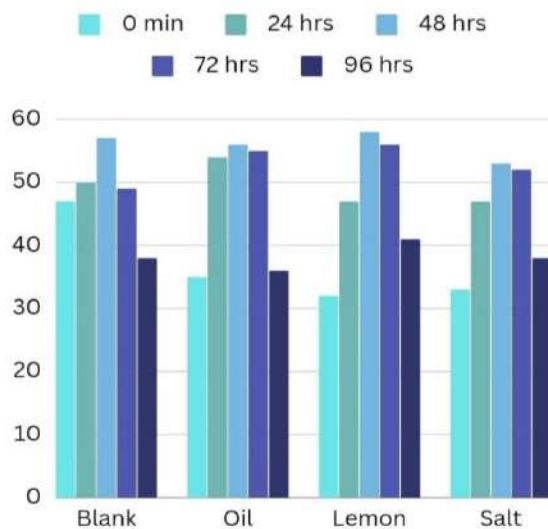
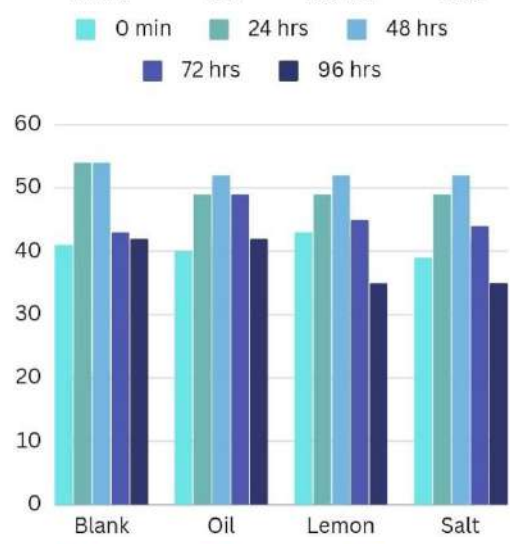
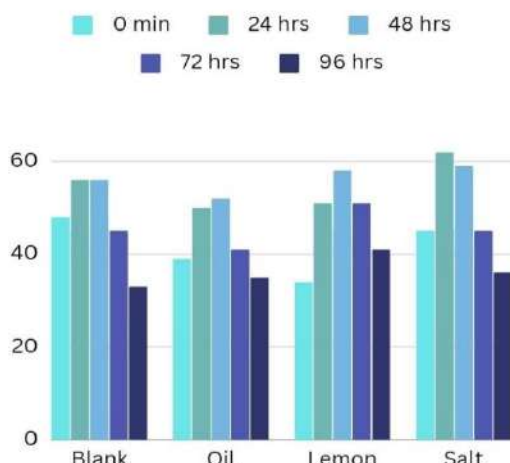
Darkness of henna on person 1

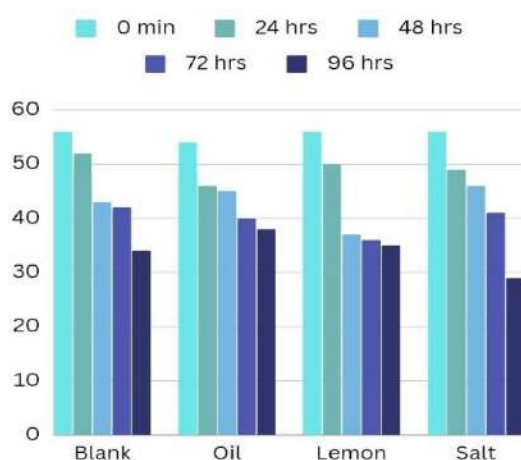


Henna tattoo



Darkness of henna on person 2-





RESULT

Dyeing of cotton fabric - From graph 1 & 2, oil gave the darkest stain, From graph 3, it's observable that 31 hours dyeing gave darker results with 31 hours oil dyeing. However, it was observed from the fig. No. 3&4, that the dyeing of henna was uneven which could be due to presence of water alongwith oil in the dye sample with oil. As oil and water are insoluble, dyeing of henna was more effective with oil and while the henna dissolved with water showed lighter stain. From graph 4, in 30 mins. Dyeing blank and sugar didn't show any change in color before and after washing. Rest all showed a decrease in color by a few percent after washing. From graph 5, blank showed an increase in color after light exposure in both dyeing processes. Sugar samples in 30 mins dyeing showed no difference in color before and after light exposure. Rest all showed a decrease in color by a few percent after light exposure. From graph 6, blank from both processes, salt and oil from 31 hrs dyeing showed no difference in color. Rest all showed a decrease in color by a few percent after rubbing. Henna tattoo - From Person 1 graph, out of three readings two graphs showed darkest staining after 24 hrs of removal. Lemon showed the highest darkness percentage. From person 2 graphs, out of the three readings two graphs showed darkest color after 48 hrs of removal. Blank showed the highest darkness

percentage. From Person 3 graphs, the darkest color was observed after removal of henna. Darkest staining was of salt, lemon and blank.

CONCLUSION

In dyeing of fabric, among oil, salt, lemon juice and sugar solution, when oil as a solvent used has darker stains in comparison to others. The fastness properties showed satisfactory results. However, further studies can be done for even staining of henna dye with oil. In hand tattoos, different people get different stains due to factors such as body temperature, hormonal changes and biochemistry of the human body.

REFERENCES

1. Slama, H. B., Chenari Bouket, A., Pourhassan, Z., Alenezi, F. N., Silini, A., Cherif-Silini, H., ... & Belbahri, L. (2021). Diversity of synthetic dyes from textile industries, discharge impacts and treatment methods. *Applied Sciences*, 11(14), 6255. <https://doi.org/10.3390/app11146255>
2. Golka, K., Kopps, S., & Myslak, Z. W. (2004). Carcinogenicity of azo colorants: influence of solubility and bioavailability. *Toxicology letters*, 151(1), 203-210. <https://doi.org/10.1016/j.toxlet.2003.11.016>
3. Ikram, M., Naeem, M., Zahoor, M., Hanafiah, M. M., Oyekanmi, A. A., Ullah, R., ... & Gulfam, N. (2022). Biological degradation of the azo dye basic orange 2 by *Escherichia coli*:



- A sustainable and ecofriendly approach for the treatment of textile wastewater. *Water*, 14(13), 2063. <https://doi.org/10.3390/w14132063>
4. Bhuiyan M.A.R., Islam A., Ali A., Islam M.N. (2017). Color and chemical constitution of natural dye henna (*Lawsonia inermis* L) and its application in the coloration of textiles, *Journal of Cleaner Production*, 167 , pp. 14-22. <https://doi.org/10.1016/j.jclepro.2017.08.142>
 5. Pasha, M. A., Anebouselvy, K., & Ramachary, D. B. (2022). Lawsonsone as synthon in the catalytic asymmetric reactions. *Tetrahedron*, 117, 132793. <https://doi.org/10.1016/j.tet.2022.132793>
 6. Mahkam, M., Nabati, M., & Rahbar Kafshboran, H. (2014). Isolation, identification and characterization of lawsonsone from henna leaves powder with soxhlet technique. *Iranian chemical communication*, 2(1, pp. 1-81, Serial No. 2), 34-38. https://scholar.google.com/scholar?start=10&q=lawsonsone+absorbs+UV+light&hl=en&as_sdt=0,5#d=gs_qabs&t=1704392339798&u=%23p%3D9jjXCIP4LXkJ
 7. Iqbal, J., Bhatti, I. A., & Adeel, S. (2008). Effect of UV radiation on dyeing of cotton fabric with extracts of henna leaves. *Indian Journal of Fibre & Textile Research*, Vol. 33, pp. 157-162. https://scholar.google.com/scholar?hl=en&as_sdt=0%2C5&q=Effect+of+UV+radiation+on+dyeing+of+cotton+fabric+with+extracts+of+henna+leaves&btnG=#d=gs_qabs&t=1704394553356&u=%23p%3Dqk76bToA65EJ
 8. Alam, M. M., Rahman, M. L., & Haque, M. Z. (2007). Extraction of henna leaf dye and its dyeing effects on textile fibre. *Bangladesh Journal of Scientific and Industrial Research*, 42(2), 217-222. https://scholar.google.com/scholar?hl=en&as_sdt=0%2C5&q=Extraction+of+henna+leaf+dye+and+its+dyeing+effects+on+textile+fibre&btnG=#d=gs_qabs&t=1704129756256&u=%23p%3DznIbI_W8EPkJ
 9. Dutta, P., Rabbi, M. R., Sufian, M. A., Mahjebin, S., & Imran, I. H. (2021). Dye ability of henna dye towards cotton fabrics in comparison with reactive dye by following reactive dyeing procedure. *Engineering and Applied Science Letters*, 4(2), 29-35. doi:10.30538/psrp-easl2021.0068 https://scholar.google.com/scholar?hl=en&as_sdt=0%2C5&q=Dye+ability+of+henna+dye+towards+cotton+fabrics+in+comparison+with+reactive+dye+by+following+reactive+dyeing+procedure&btnG=#d=gs_qabs&t=1704129756256&u=%23p%3DznIbI_W8EPkJ
 10. Udeani, N. A. (2015). Potential of Henna leaves as dye and its fastness properties on fabric. *International Journal of Chemical and Molecular Engineering*, 9(12), 1459-1466. <https://doi.org/10.5281/zenodo.1110762>
 11. Rabia, S. A., Samad, B. A., Mazhar, H. P., & Alvira, A. A. (2019). An efficient ultrasonic and microwave assisted extraction of organic Henna dye for dyeing of synthetic polyester fabric for superior color strength properties. *Ind Text*, 70, 303-308.
 12. https://scholar.google.co.in/scholar?q=effect+of+henna+dyeing+at+different+pH+research+papers&hl=en&as_sdt=0&as_vis=1&oi=scholar#d=gs_qabs&t=1685550166342&u=%23p%3DIRutaVs8vmAJ
 13. Yusuf, M., Shahid, M., Khan, M. I., Khan, S. A., Khan, M. A., & Mohammad, F. (2015). Dyeing studies with henna and madder: A research on effect of tin (II) chloride mordant. *Journal of Saudi Chemical Society*, 19(1), 64-72.



14. https://scholar.google.co.in/scholar?q=fastness+properties+of+henna+dyeing+research+papers&hl=en&as_sdt=0&as_vis=1&oi=scholar#d=gs_qabs&t=1685550703829&u=%23p%3D8UuIn9ucr34J
15. Yusuf, M., Ahmad, A., Shahid, M., Khan, M. I., Khan, S. A., Manzoor, N., & Mohammad, F. (2012). Assessment of colorimetric, antibacterial and antifungal properties of woollen yarn dyed with the extract of the leaves of henna (*Lawsonia inermis*). *Journal of cleaner production*, 27, 42-50.
16. https://scholar.google.co.in/scholar?q=fastness+properties+of+henna+dyeing+research+papers&hl=en&as_sdt=0&as_vis=1&oi=scholar#d=gs_qabs&t=1685550846509&u=%23p%3Dgcdd0eIa4FEJ
17. Mirjalili, M., Nazarpour, K., & Karimi, L. (2011). Eco-friendly dyeing of wool using natural dye from weld as co-partner with synthetic dye. *Journal of Cleaner Production*, 19(9-10), 1045-1051. <https://doi.org/10.1016/j.jclepro.2011.02.001>
18. Miraj, S., & Kiani, S. A. (2016). Review study of chemical constituents and side-effects of black henna for children. *Der Pharmacia Lettre*, 8(4), 277-281. https://scholar.google.com/scholar?hl=en&as_sdt=0%2C5&q=Review+study+of+chemical+constituents+and+side%20%80%91effects+of+black+henna+for+children&btnG=#d=gs_qabs&t=1704129813025&u=%23p%3D5aDqm5dbIK4J
19. Affat, S. S. (2021). Classifications, advantages, disadvantages, toxicity effects of natural and synthetic dyes: A review. *University of Thi-Qar Journal of Science*, 8(1), 130-135. <http://doi.org/10.32792/utq/utjsci/v8/1/21>
20. Rahman Bhuiyan, M.A., Ali, A., Islam, A. et al. (2018) Coloration of polyester fiber with natural dye henna (*Lawsonia inermis* L.) without using mordant: a new approach towards a cleaner production. *Fash Text* 5, 2. <https://doi.org/10.1186/s40691-017-0121-1>
21. Kazandjieva, J., Grozdev, I., & Tsankov, N. (2007). Temporary henna tattoos. *Clinics in dermatology*, 25(4), 383-387. <https://doi.org/10.1016/j.clindermatol.2007.05.013>
22. Rubio, L., Lores, M., & Garcia-Jares, C. (2020). Monitoring of natural pigments in henna and jagua tattoos for fake detection. *Cosmetics*, 7(4), 74. <https://doi.org/10.3390/cosmetics7040074>
23. Jovanovic, D. L., & SLAVKOVIC-JOVANOVIC, M. R. (2009). Allergic contact dermatitis from temporary henna tattoo. *The Journal of Dermatology*, 36(1), 63-65. <https://doi.org/10.1111/j.1346-8138.2008.00588.x>

HOW TO CITE: Vibha S. Bhagat, Afsana Mondal, Mahedaraksha Junaid Ahmed, Juliet Miranda, Henna Dyeing On Cotton Fabric With Different Natural Mordants And Henna Tattoo With Different Natural Solvents, *Int. J. of Pharm. Sci.*, 2024, Vol 2, Issue 7, 1271-1284. <https://doi.org/10.5281/zenodo.12761219>

